

Chapter 3 Chemical Building Blocks Of Life Isomers

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[Chemical Building Blocks Chapter 3. STUDY. PLAY. Alkali Metal. An element in Group 1 of the periodic table. Alkaline Earth Metal. An element in Group 2 of the periodic table. Alloy. A mixture of two or more metals. Actinide. An element in the second row of the rare earth elements in the periodic table.](#)

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[Chapter 3 The Chemical Building Blocks of Life 1 What are ...](#)

[Atoms: The Building Blocks of Matter An atom is the smallest particle of an element that retains the chemical properties of that element. CHAPTER 3 STM Image of Impure Gold Surface](#)

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[Chapter 3: Atoms, The Building Blocks of Matter. Ms. Douglas. Chemistry. ... Describe which chemical laws can be explained by Dalton 's theory. Think critically. Three compounds containing potassium \(K\) and oxygen \(O\) are compared. Analysis shows that, for each 1.00 g of O, the compounds have 1.22 g, 2.44 g, and 4.89 g of K. **respectively. ...](#)

[Chapter 3: Atoms, The Building Blocks of Matter](#)

[Chemistry Chapter 3. 1. Chapter 3 Atoms: the Building Blocks of Matter. 2. --The parts that make up an atom are called subatomic particles . ----1\) Protons \(p + \) positively charged particle ----2\) Neutron \(n o \) neutral particle \(uncharged\) ----3\) Electrons \(e - \) negatively charged particle ----Neutrons and Protons are located in the nucleus of an atom. ---Electrons orbit around the nucleus.](#)

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[Chapter Contents 3.1 Carbon: The Framework of Biological Molecules 3.2 Carbohydrates: Energy Storage and Structural Molecules 3.3 Nucleic Acids: Information Molecules 3.4 Proteins: Molecules with Diverse Structures and Functions 3.5 Lipids: Hydrophobic Molecules The Chemical Building Blocks of Life CHAPTER 3 rav68132_ch03_033-058.indd 33 11/24/15 8:01 PM](#)

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[Chapter 3 - Chemistry](#)

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[Chapter 3 - The Chemical Building Blocks of Life - Review ...](#)

[CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table. However, no single atom of boron has a mass of exactly 10.81 amu. How can you explain this difference?](#)

[3 Atoms: The Building Blocks of Matter](#)

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[Chapter 3 - The Chemical Building Blocks of Life - Review Questions - Synthesize - Page 58: 2 Answer Hydrogen bonds are most important for hydrophobic macromolecules: carbohydrates, nucleic acids, and proteins.](#)

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